*	Chinese English Kinetics of Materials					
Course Name						
* Credits	3		* Teaching Hours	48 1 =16		
* Semester	Sprin	g	* Cross-semester?	No Spanning over Semesters		
* Course Type	Program Core Course		* Course Type	For full-time students		
* Course Category	Specialized Course		Targeting Students	All graduates		
* Instruction Language	English		Teaching Method	In class teaching		
* Grade	Letter g	grading	Exam Method	Written Exam		
* School						
Subject						
	Name	ID	School	E-mail		
Person in charge				hongwang2@sjtu.edu.cn		
* ( ) Course Description	1) 2) 3) 4)			200		
* English Course Description	This course provides a foundation for the advanced understanding of the phenomenological and atomistic kinetic process in materials. It emphasizes comprehension of fundamental concepts and stresses on development of students' ability of quantitative analysis. The course starts from a brief review of classical thermodynamics necessary for understanding of phase diagrams, followed by some key concepts such as flux and driving force. The principles of chemical reaction kinetics will be introduced, as well as several important applications such as in thin film growth process and chemical vapor deposition. The center stage of the course is given to the diffusion process in solid materials as well as the analytical and numerical methods to solve diffusion problems. Then 1/4					

	the kinetic process driven by capillarity force will be discussed, followed by nucleation and growth of crystals and the phase transformation. Several important metallurgical processes such						
	as solidification, spinodal decomposition, etc. will be discussed. Before the end of the course, the						
	students will be exposed to basics of computer simulation methods and software for thermodynamics and kinetics.						
	The overall goals of this course are to:						
	1) develop an understanding of why materials and microstructures undergo changes by reinforcing						
	and significantly extending concepts introduced in thermodynamics courses:						
	microstructure of materials by discussing mechanisms and rates of diffusion and the role of						
	driving force on diffusional processes; 3) discuss a variety of phase transformations phenomena and the effects of temperature and						
	driving force on the nature of the transformation and its impact on the resulting microstructure:						
	4) introduce the methods and software for thermodynamic and kinetic simulation, which became more and more important in today's materials design and development:						
	In short, the course will give the students the tools required to understand how and why phase						
	transformations occur, and how and why microstructures can be controlled and developed.						
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	The class will be delivered in 24 lectures, where each lecture contains two	45min classes.
	I INTRODUCTION AND BACKGROUND Introduction thermodynamics Phase diagrams, Driving force, flux	1 lecture 1 lecture
* English Syllabus	II KINETICS OF CHEMICAL REACTIONS Chemical reaction kinetics, adsorption isotherms Thin film growth Rate controlling steps; CVD	1 lecture 1 lecture
	<ul> <li>III DIFFUSION IN SOLIDS</li> <li>Fick's Laws and solutions to Fick's laws</li> <li>Interdiffusion</li> <li>Self, tracer, intrinsic and interdiffusion coefficients</li> <li>Atomistic models of diffusion, Diffusion in ionic crystals</li> <li>Multipath imperfections</li> </ul>	2 lecture 1 lecture 1 lecture 1 lecture 1 lecture
	IV KINETICS DRIVEN BY CAPILLARITY FORCES Capillarity forces on surfaces, grain growth Surface energy anisotropy Particle coarsening, sintering	2 lectures 1 lecture 1 lecture
	V KINETICS OF PHASE TRANSFORMATIONS Nucleation and growth Solidification Order-disorder Reactions Spinodal decomposition Martensitic transformation	2 lectures 1 lecture 1 lecture 1 lecture 1 lecture
	VI MODELLING OF KINETIC PROCESS Computational thermodynamics (CALPHAD) Diffusion simulation, Phase field simulation Computational lab	1 lecture 2 lecture 1 lecture
	Final Exam	
* Requirements	$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	
* English Requirements	<ol> <li>10 Homework assignments, 2% each, 20%</li> <li>Midterm Exam, covers first 10 lectures 30% each</li> <li>Final Exam, covers later 9 lecture, 30%</li> <li>Participation: attendance and class discussion, 10%</li> <li>Quiz: simple questions due next morning 6:00am after each class, 5%</li> <li>Project: computation lab work, 5%</li> <li>The final grade will be calculated based on the sum of the above.</li> </ol>	6
* Resources	: KINETICS IN MATERIALS SCIENCE AND ENGINEERING CRC Press, 2017	Dennis W. Readey,